

Plattsburgh AFB Case Study: Influence of Modeling Fe(II) Transport on Redox Zonation

Prepared by
Grant R. Carey
Porewater Solutions, Ottawa, Ontario

Prepared for
Dr. Brian Looney
Savannah River National Laboratory (SRNL)

January 13, 2007

Figure 1 – Site Map

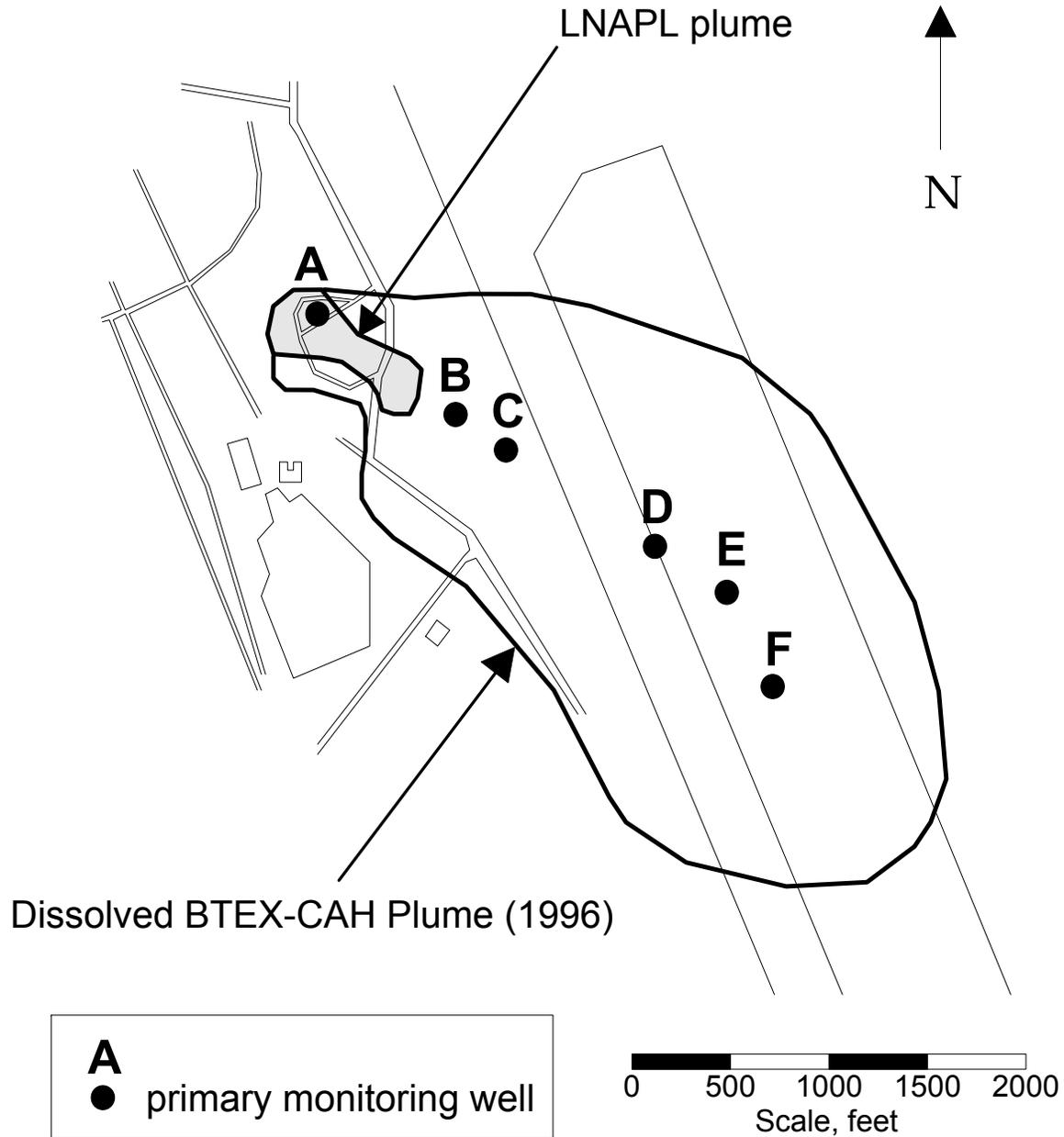


Figure 2 – Radial Diagram Map: VOCs (1995)

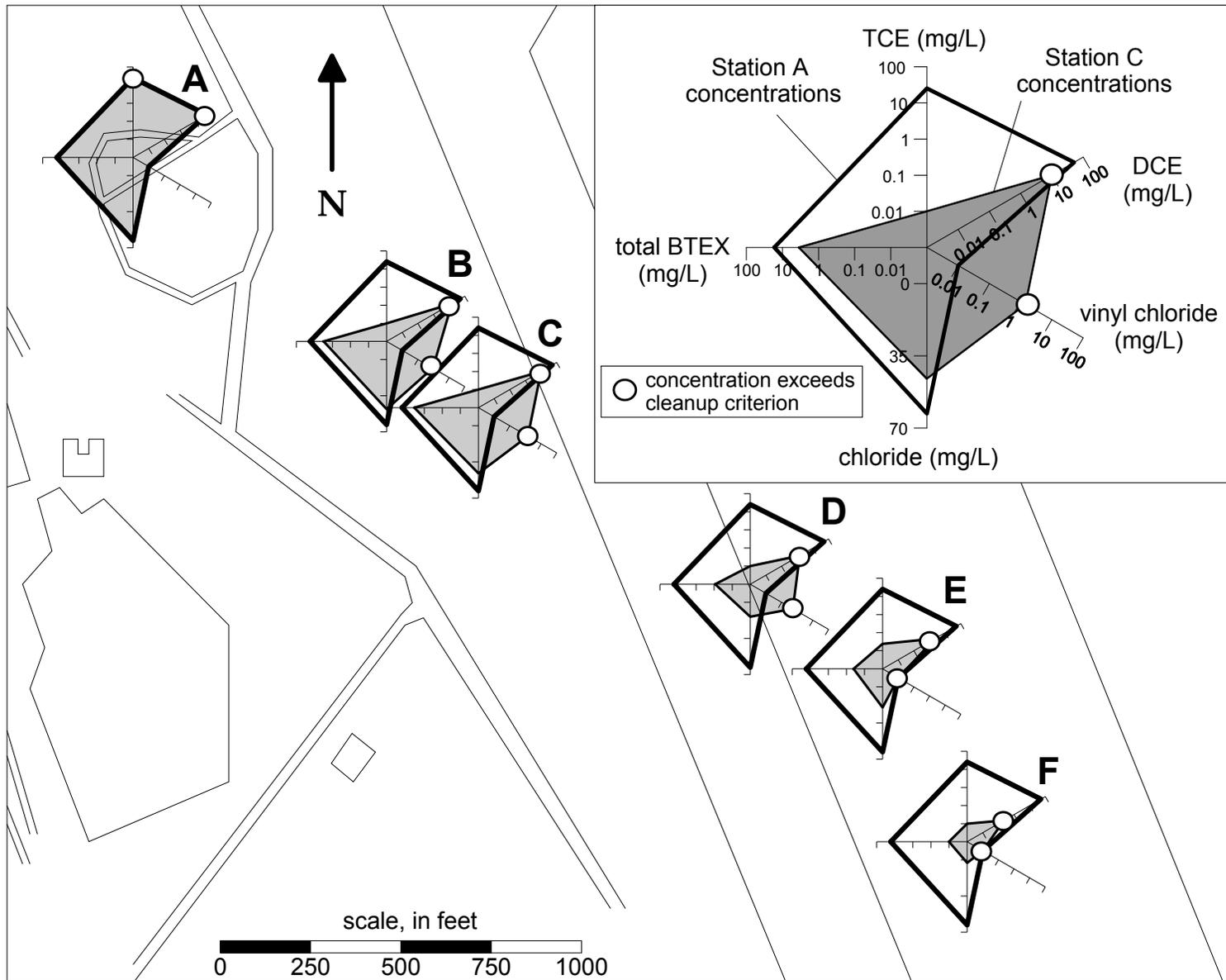


Figure 3 – Radial Diagram Map: Redox Indicators (1995)

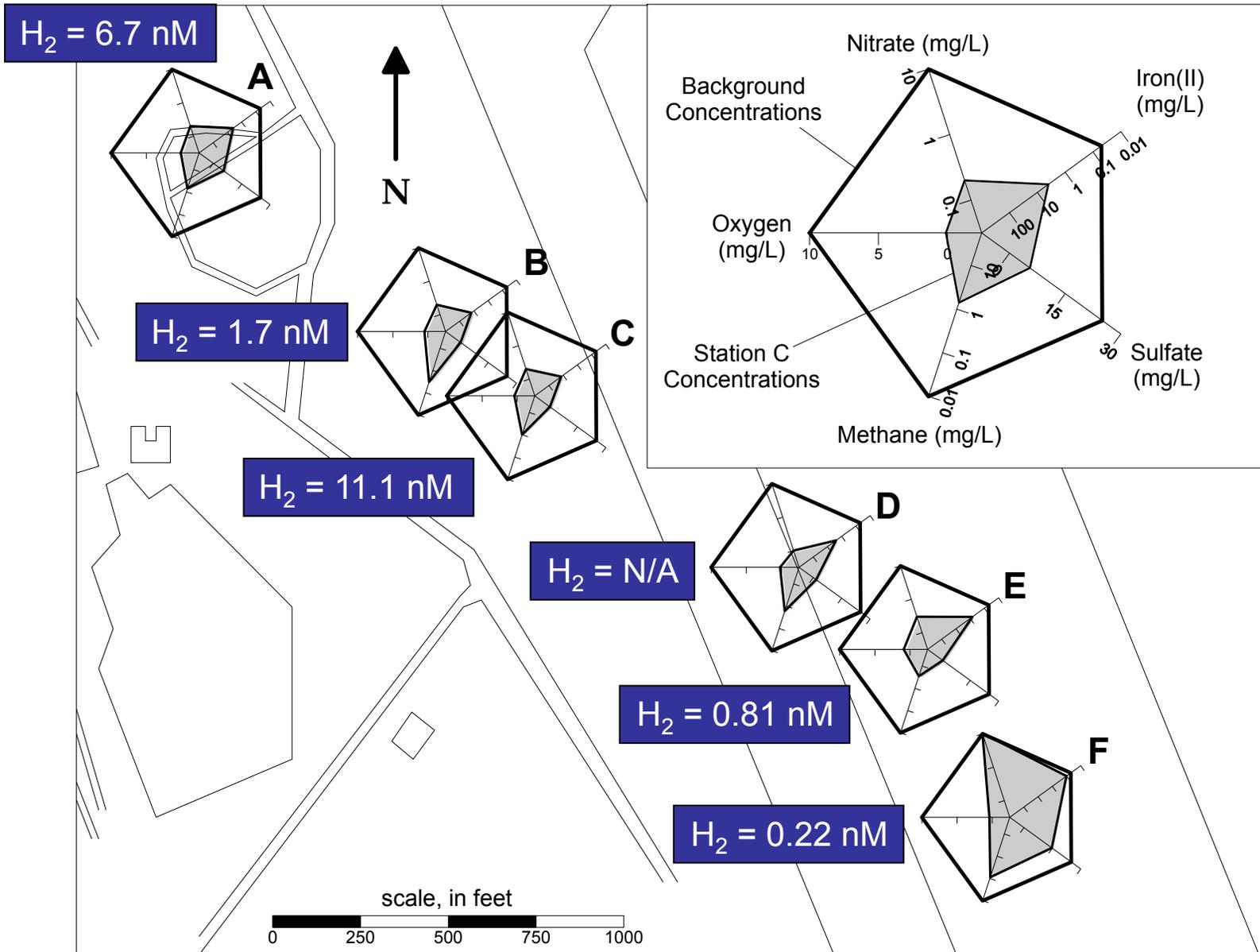


Figure 4 – Transport Model Domain and Boundary Conditions

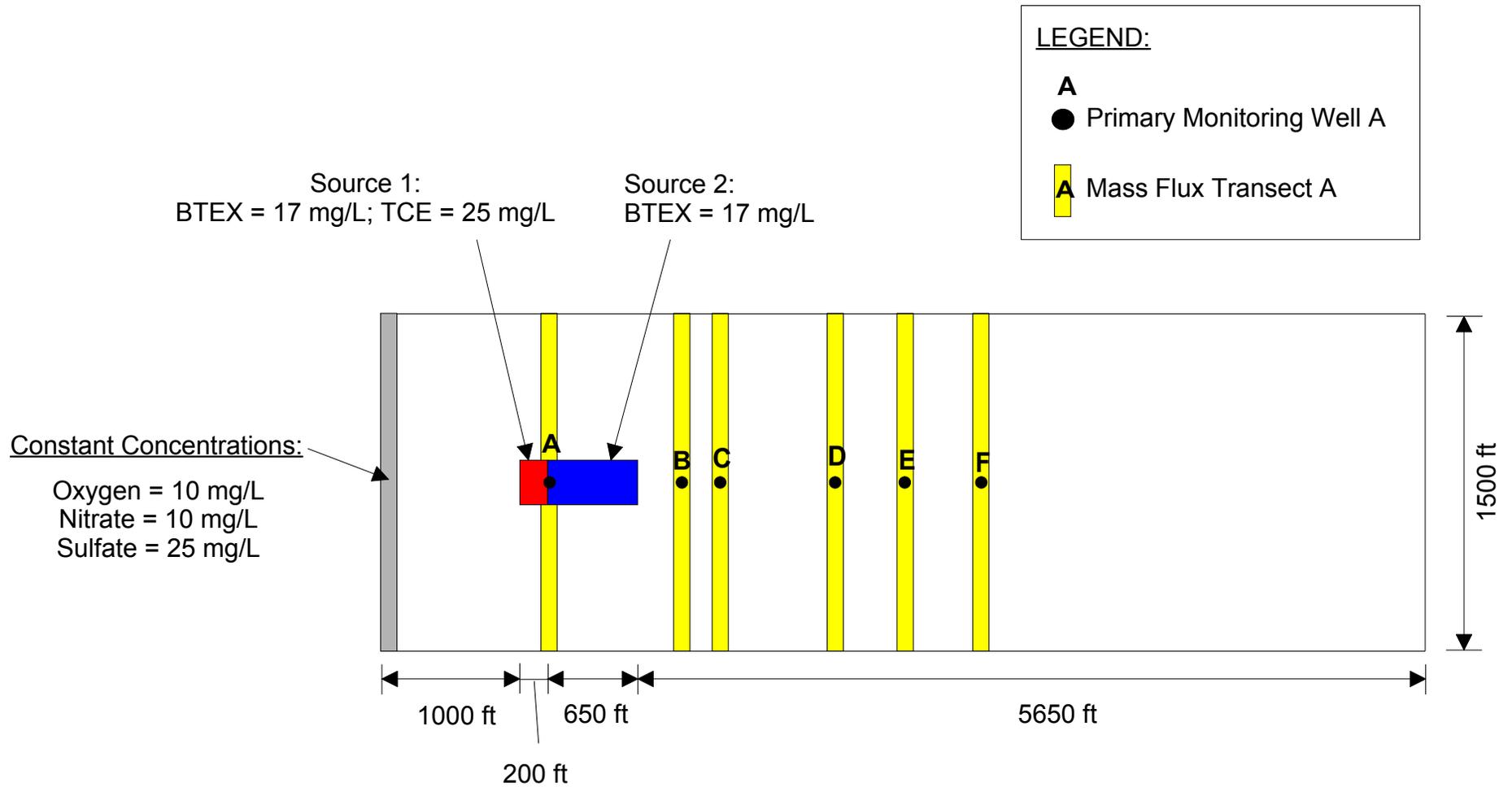


Figure 5 – Calibrated Half-Lives (years)

	CH ₄	SO ₄	Fe ³⁺	NO ₃	O ₂	
BTEX	0.6	0.6	0.6	0.6	INST.	
TCE	0.2	0.3	1		0.25	
DCE	2	2	2		0.25	0.2
VC	0.2		0.02/0.2*		0.25	0.2
Fe ²⁺					INST.	
CH ₄	2	2	2	2	0.2	

LEGEND:

	<i>Mechanism</i>	<i>Kinetics</i>
■	Oxidation	Instantaneous
■	Oxidation	First-Order
■	CH ₄ Co-metabolism	First-Order
■	Dehalogenation	First-Order
□	No Degradation	n/a

Figure 6a – Model vs. Field Concentrations: VOCs

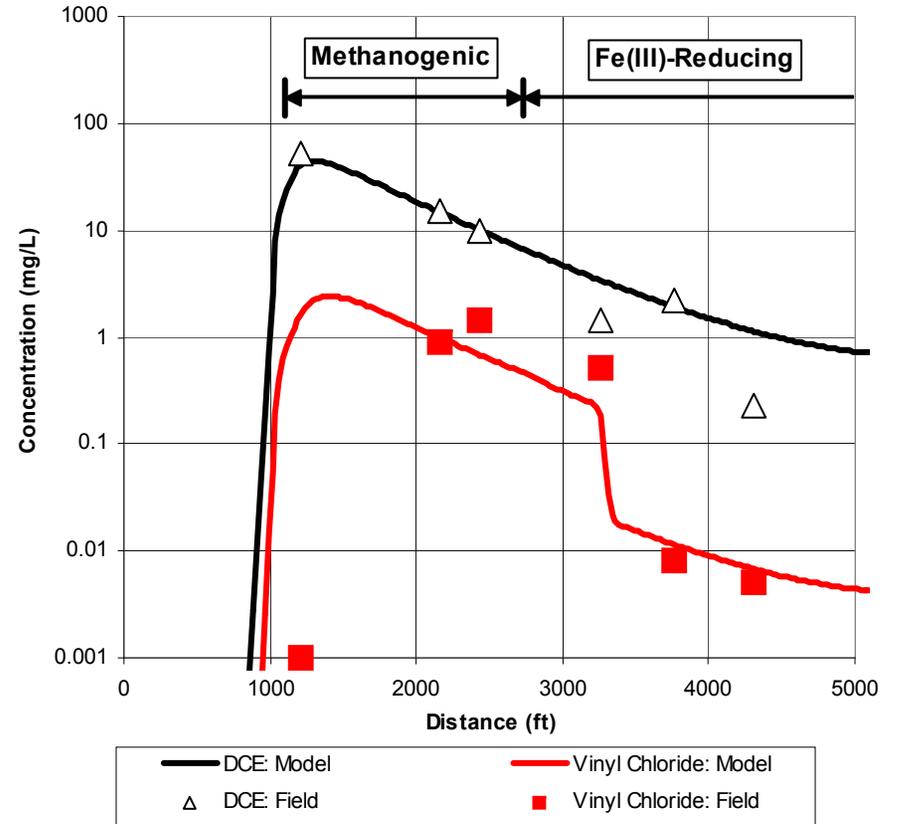
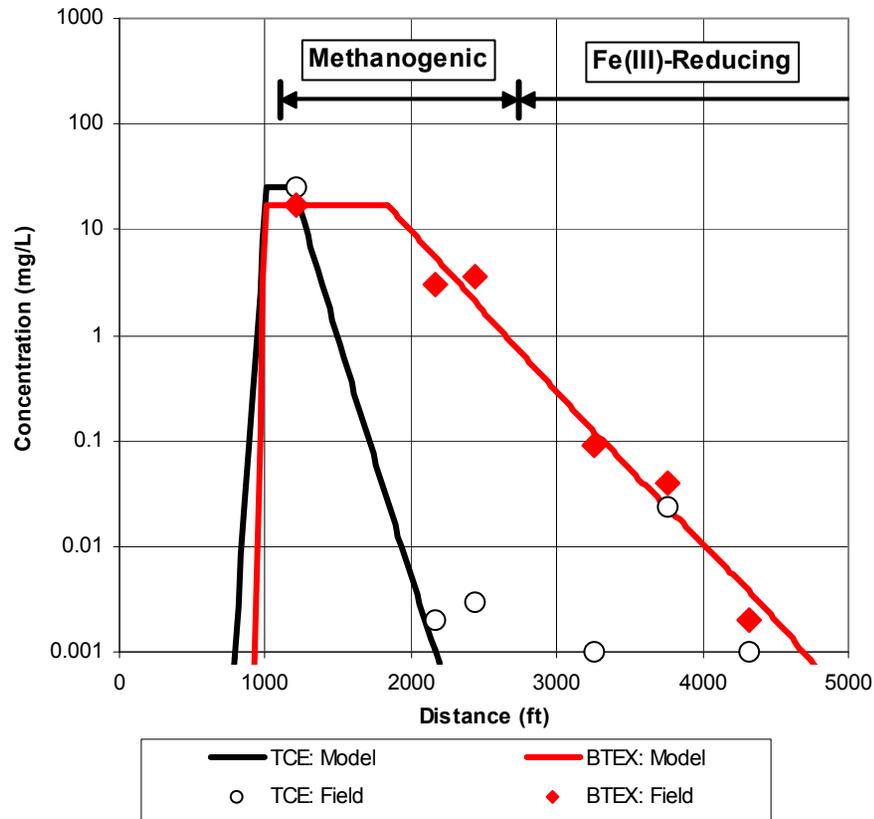


Figure 6b – Model vs. Field Concentrations: Redox Indicators

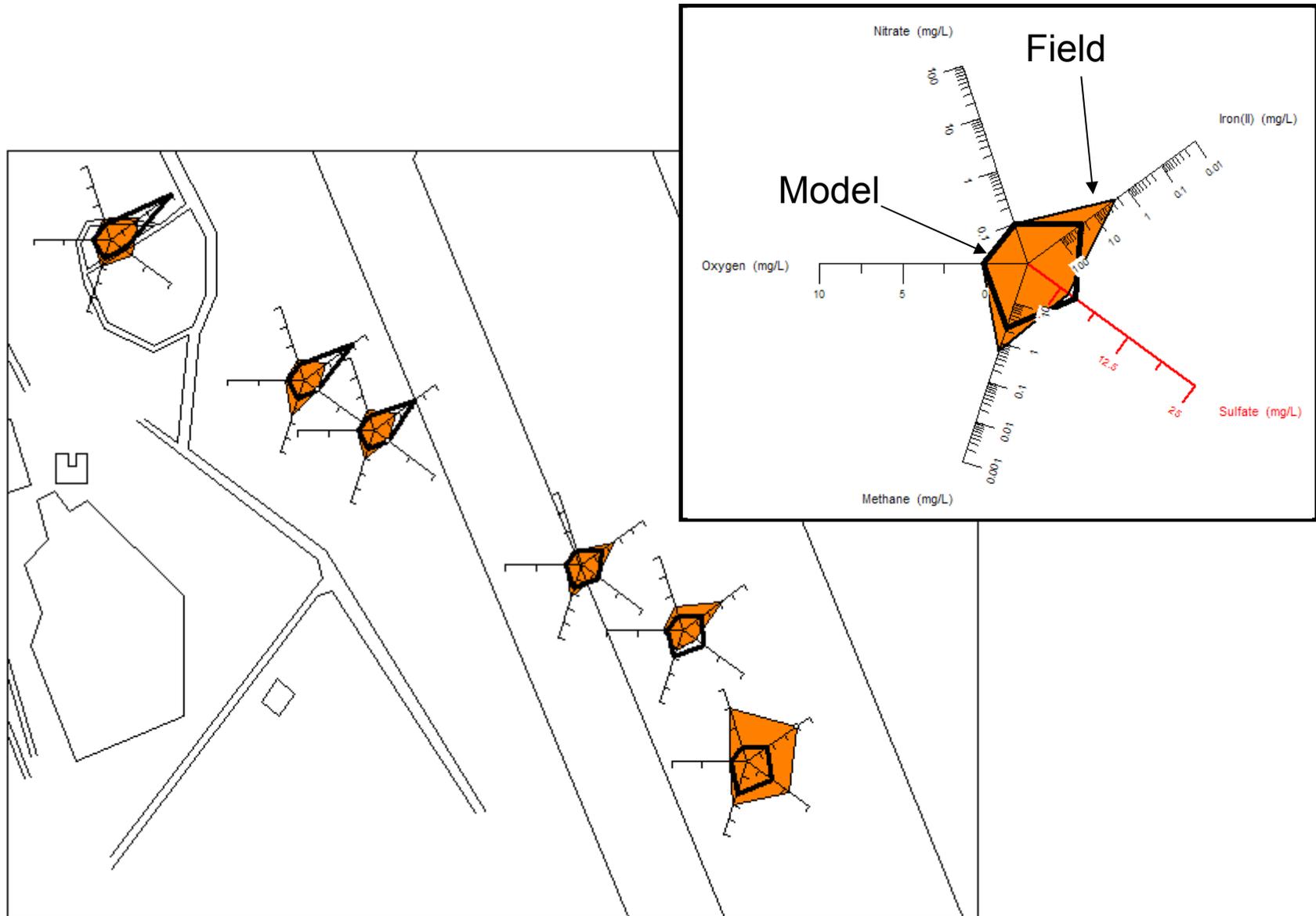


Figure 7 – Simulated TCE and Redox Zones Distributions over Time

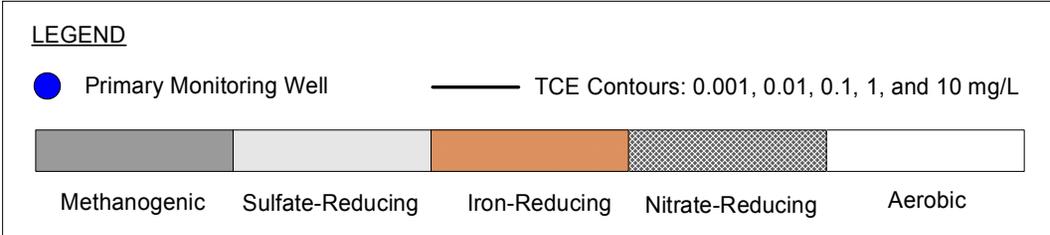
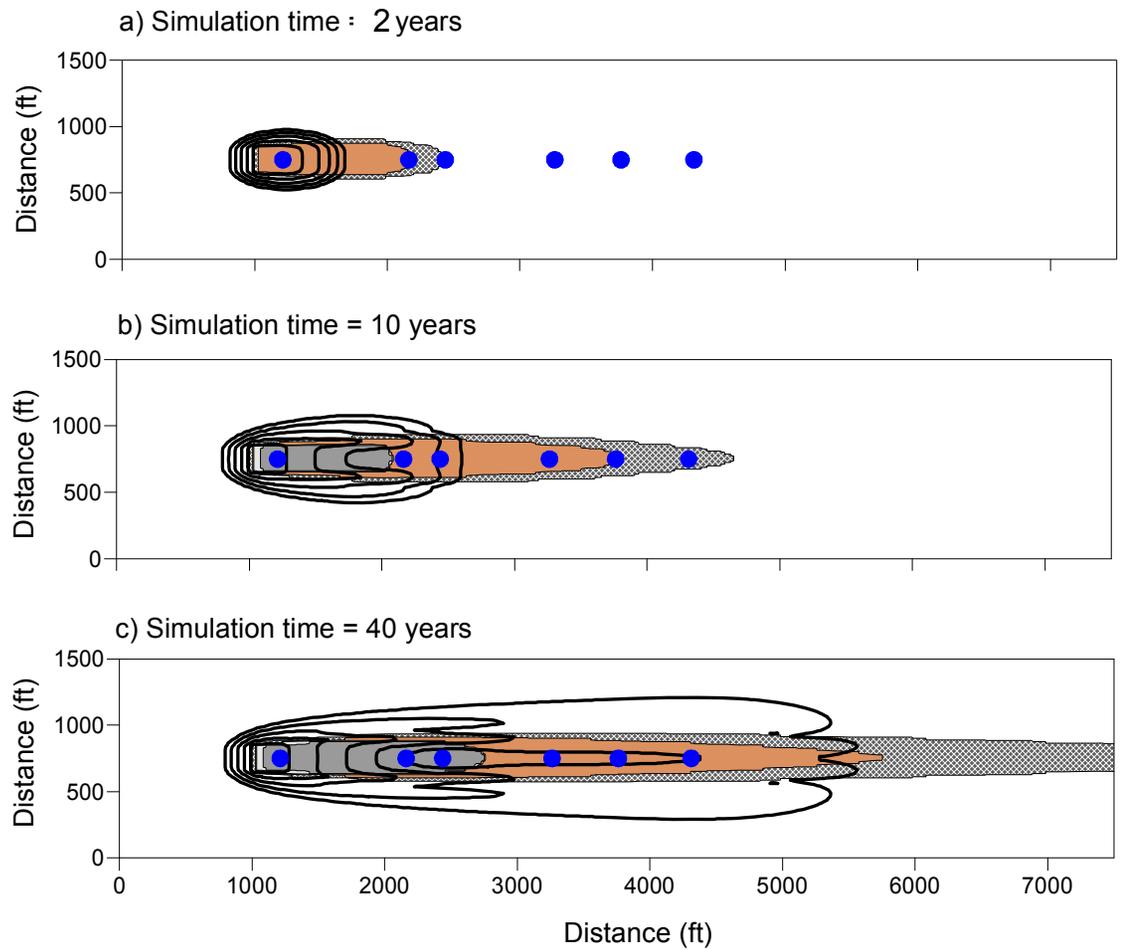
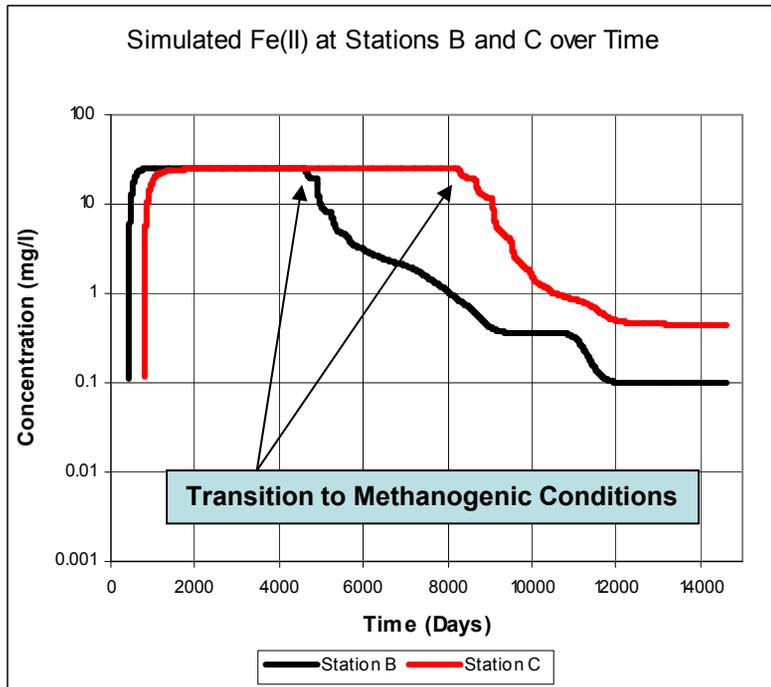


Figure 8 – Simulated Iron Concentrations

8.a) Simulated Fe(II) vs. time at Stations B and C



8.b) Simulated Fe(III) vs. time at Stations A, C, and D

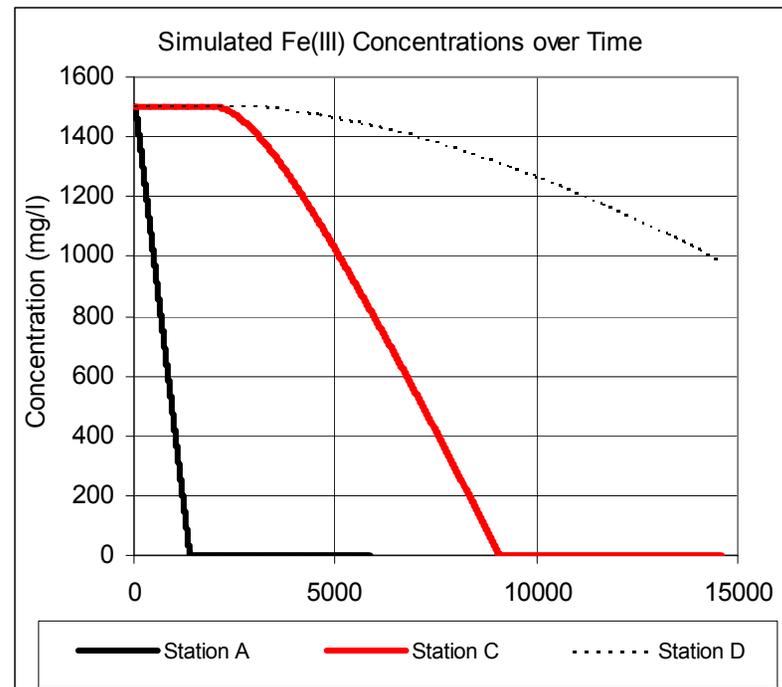
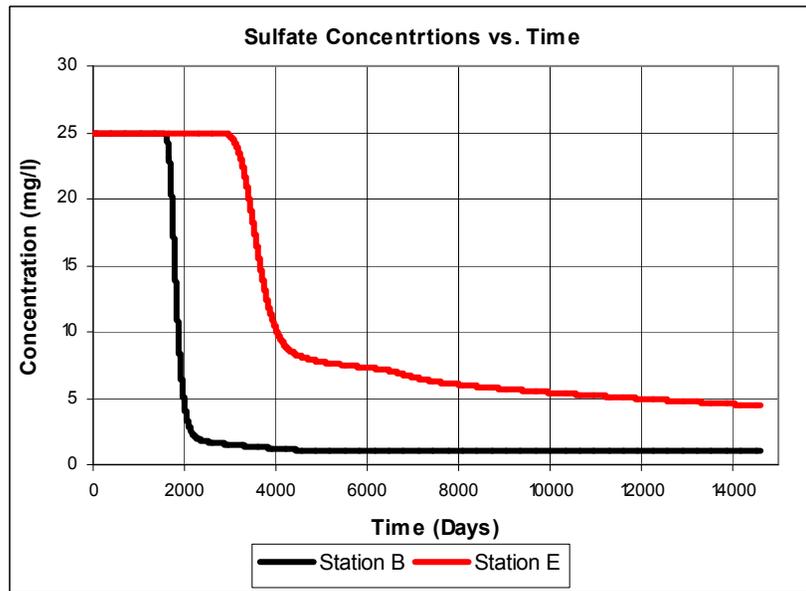


Figure 9 – Simulated Sulfate and Methane Concentrations

8.a) Simulated Sulfate vs. time at Stations B and E



8.a) Simulated Methane vs. time at Stations A, C, and E

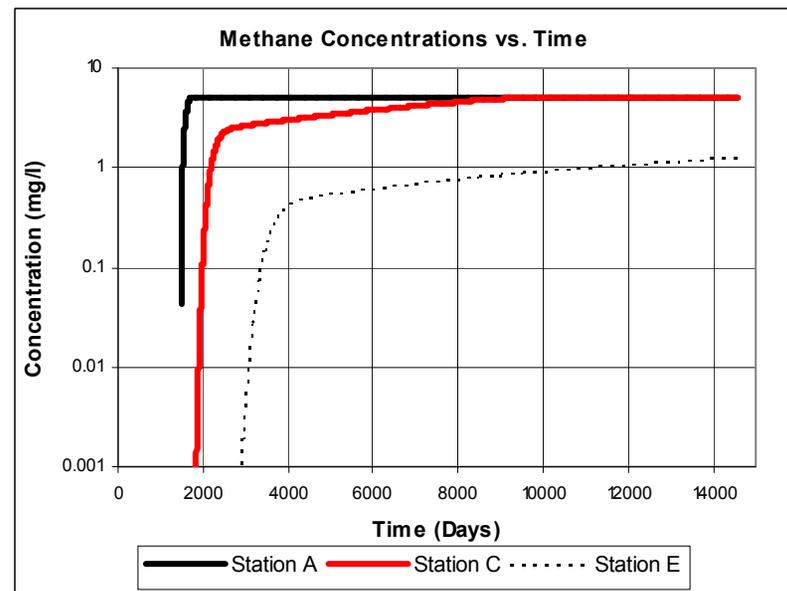


Figure 10 – Simulated Vinyl Chloride Concentrations

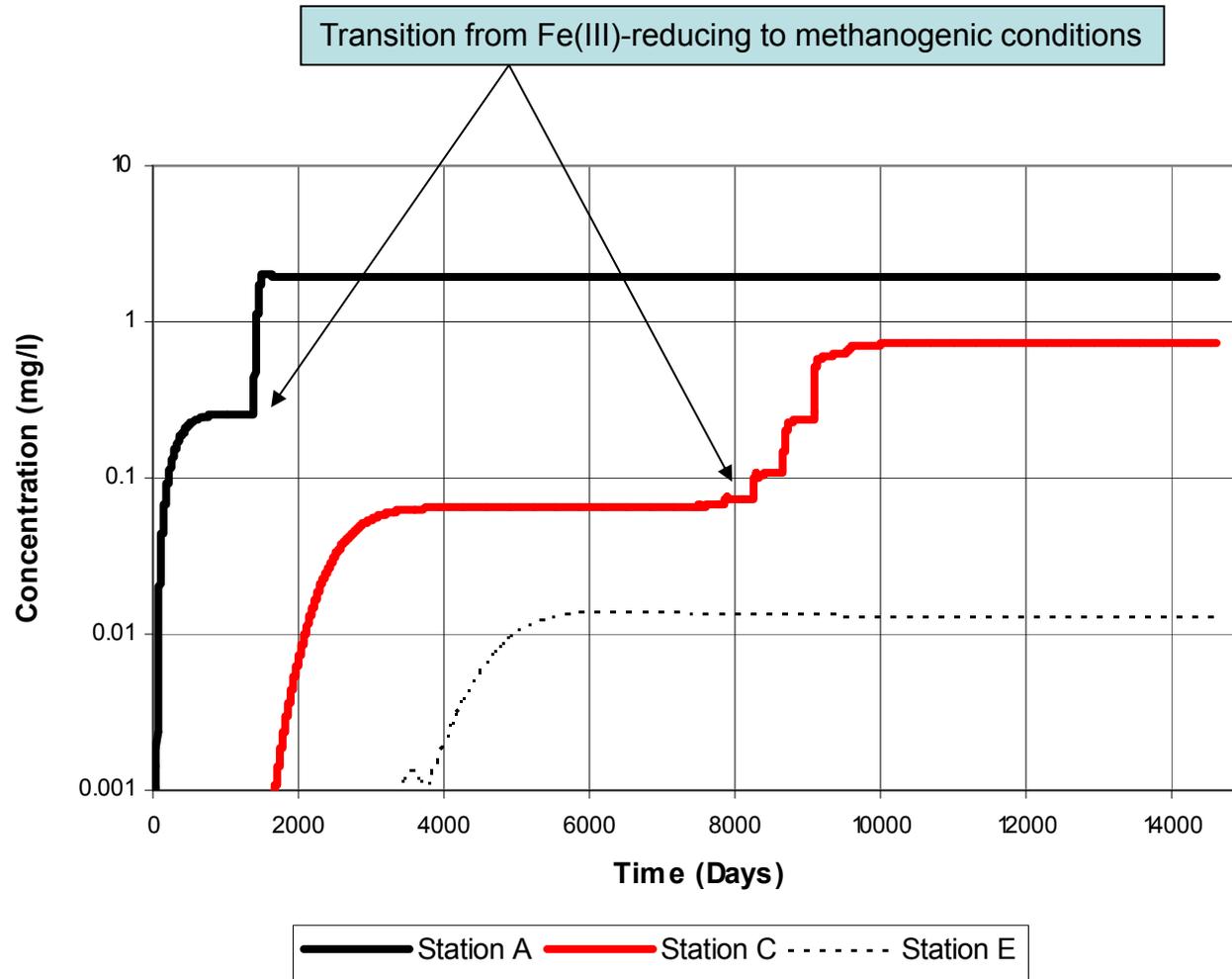
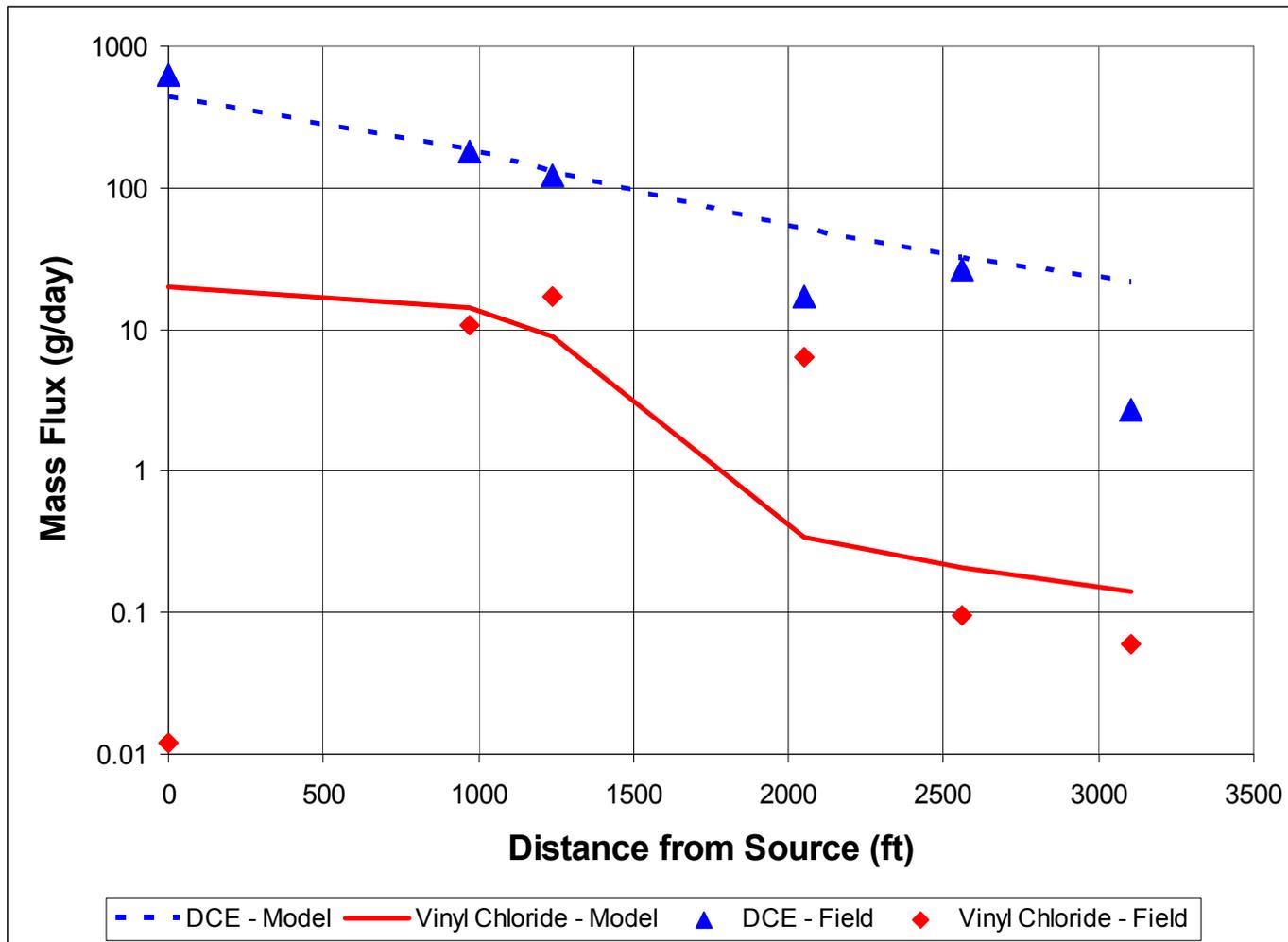


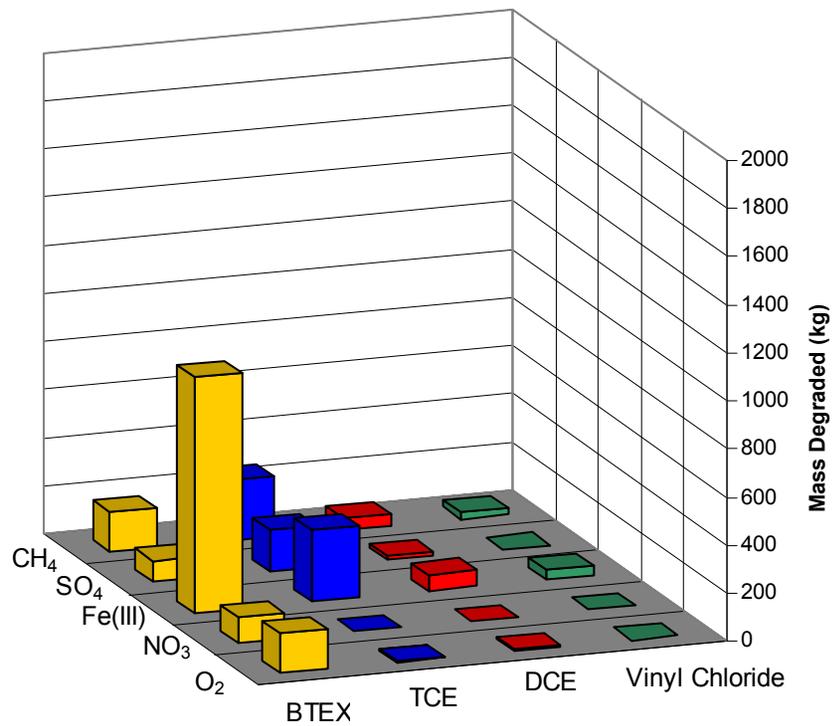
Figure 11 – Mass Flux Comparison: Model vs. Field (t = 40 years)



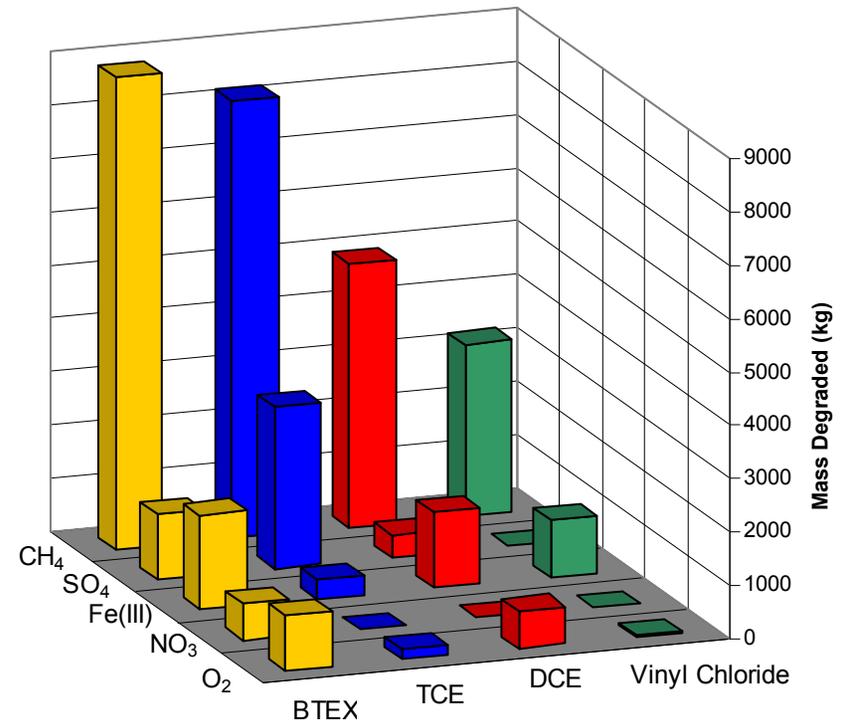
Note – field-estimated mass flux is based on concentrations at key monitoring wells along plume centerline, and assumption that main plume width is equal to source width of 200 feet.

Figure 12 – Simulated Biodegradation Mass Balance (t=30 years)

12.a) Simulated Mass Degraded by Redox Zone
(t = 5 years)



12.a) Simulated Mass Degraded by Redox Zone
(t = 40 years)



Redox Zone Legend: O₂ = aerobic; NO₃ = nitrate-reducing;
Fe(III) = iron-reducing; SO₄ = sulfate-reducing;
CH₄ = methanogenic.

Figure 13 – Simulated Influence of Coupled Fe(II)-O₂ Reaction

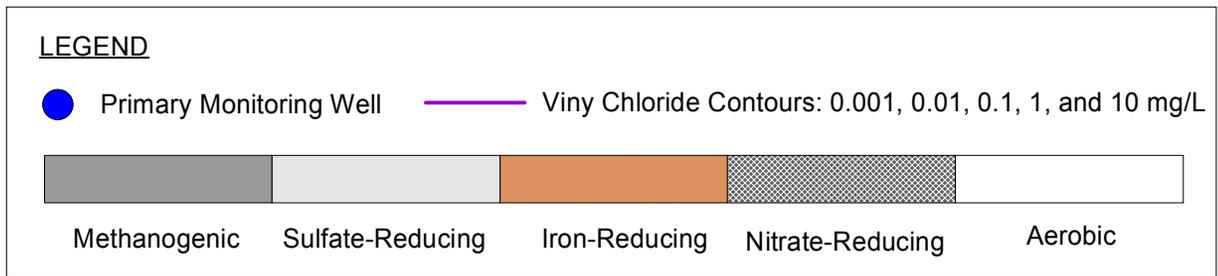
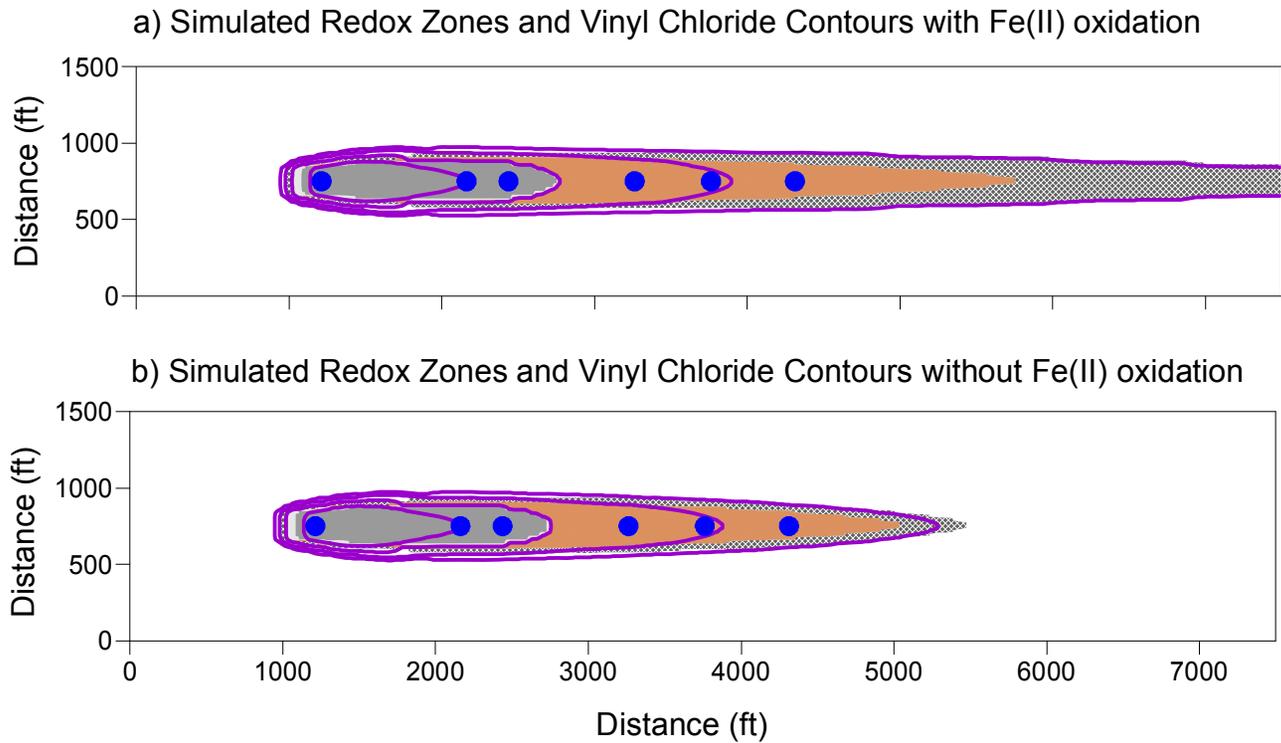


Figure 14 – Influence of Fe(II) Oxidation on Vinyl Chloride Plume Extent

